

BOAT-CHAIR CONFORMATIONAL PREFERENCE OF EXO,EXO-9-OXOBICYCLO[3.3.1]NONANE-2,4-DICARBOXYLIC ACID

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Abstract. The ^1H and ^{13}C nmr spectra of exo,exo-9-oxobicyclo[3.3.1]nonane-2,4-dicarboxylic acid, 3, in DMSO-d_6 or alkaline D_2O , clearly show that it exists in a boat-chair-conformation with equatorial carboxyl groups, thus being the first case of boat-chair preference of a bicyclo[3.3.1]nonan-9-one due to the presence of exo,exo-2,4-substituents.

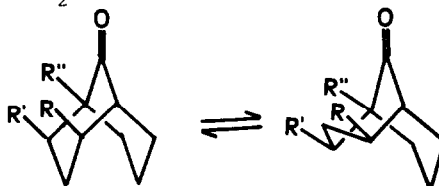
The conformational analysis of bicyclo[3.3.1]nonane derivatives is a matter of continuing interest. ¹ Bicyclo[3.3.1]nonane and bicyclo[3.3.1]nonan-9-one exist in a flattened chair-chair conformation due to the non-bonded $\text{H}_{3\text{endo}}-\text{H}_{7\text{endo}}$ interaction, in rapid equilibrium with the less stable boat-chair conformation. While introduction of an endo substituent at C_3 shifts the equilibrium entirely towards the boat-chair-conformation, ^{1c} the effect of exo-substituents at C_2 and C_4 is not so drastic. In a recent publication ^{1b} we have described that exo,exo-2,4-dimethoxybicyclo[3.3.1]nonan-9-one, 1, in CDCl_3 solution, adopts preferently a chair-chair conformation with axial methoxy groups as shown by ^1H and ^{13}C nmr spectroscopy. More recently, Baker and Frazer ² have also shown a chair-chair conformational preference for exo-2-acetoxy-4(exo),5-dimethylbicyclo[3.3.1]nonan-9-one, 2.

We describe herein that exo,exo-9-oxobicyclo[3.3.1]nonane-2,4-dicarboxylic acid, 3, prepared in a straightforward manner by KMnO_4 oxidation of anti-tricyclo[4.3.1.1^{2,5}]undec-3-en-10-one, 4 under phase transfer catalysis conditions, ⁵ exists almost exclusively in the boat-chair conformation, in DMSO-d_6 or alkaline D_2O solution.

1, $\text{R} = \text{R}' = \text{OMe}$, $\text{R}'' = \text{H}$

2, $\text{R} = \text{OAc}$, $\text{R}' = \text{R}'' = \text{Me}$

3 $\text{R} = \text{R}' = \text{COOH}$, $\text{R}'' = \text{H}$



The 200 MHz ^1H nmr spectrum of diacid 3 in DMSO-d_6 showed: a double triplet centered at δ 2.96 assigned to $\text{H}_2(\text{H}_4)$ with coupling constants $J_{\text{H}_2-\text{H}_3\text{exo}} = 13.0$ Hz and $J_{\text{H}_2-\text{H}_3\text{endo}} = J_{\text{H}_2-\text{H}_1} = 4.3$ Hz, a broad singlet at δ 2.57 assigned to $\text{H}_1(\text{H}_5)$, a double triplet at δ 2.17 assigned to $\text{H}_{3\text{endo}}$ with $J_{\text{H}_{3\text{endo}}-\text{H}_3\text{exo}} = 13.0$ Hz and $J_{\text{H}_{3\text{endo}}-\text{H}_2(4)} = 4.3$ Hz, a quartet at δ 1.23 assigned to $\text{H}_{3\text{exo}}$ with $J_{\text{H}_{3\text{exo}}-\text{H}_{3\text{endo}}} = J_{\text{H}_{3\text{exo}}-\text{H}_2(4)} = 13.0$ Hz, the rest of the protons appearing as a complex absorption at δ 1.4-2.0. On irradiation at δ 2.96, the absorptions corresponding to

$H_{3\text{exo}}$ and $H_{3\text{endo}}$ became doublets, and on irradiation at δ 2.57, the absorption corresponding to $H_2(H_4)$ became a double doublet while those corresponding to $H_{3\text{exo}}$ and $H_{3\text{endo}}$ remained unchanged. The high value of $J_{H_{3\text{exo}}-H_2(H_4)}$ is indicative of a trans-diaxial relationship between $H_{3\text{exo}}$ and $H_2(H_4)$, that can be only attained in the boat-chair conformation of 3. By contrast, the corresponding value for compound 1 is 3.90 Hz. Two other features are worth mentioning: a) the upfield shift of $H_{3\text{exo}}$ in 3 (1.23 ppm) with respect to 1 (2.10 ppm), and b) the absence of a W coupling between $H_1(H_5)$ and $H_{3\text{exo}}$ in 3, a type of coupling that was observed in 1. Both features confirm the boat-chair conformational preference of diacid 3, since in this conformation $H_{3\text{exo}}$ is very close to the π cloud of the ketonic function,⁶ and neither $H_{3\text{exo}}$ nor $H_{3\text{endo}}$ can W couple with $H_1(H_5)$.

This was additionally supported by the ^{13}C nmr spectrum³ in DMSO-d_6 , since the chemical shift for the sp^3 carbon atoms of 3 are in good agreement with those predicted taking the average values for boat-chair-bicyclo{3.3.1}nonan-9-one⁷ and correcting them for the presence of equatorial carboxyl groups at C_2 and C_4 . Specially significant is the shielding of C_7 (δ 15.7 ppm) for which a value of 15.7 ppm is predicted in the boat-chair conformation vs 21.0 ppm in the chair-chair one.

Similar conclusions can be drawn from the 200 MHz ^1H nmr spectrum of compound 3 in alkaline (NaOD) D_2O which is fully comparable³ to the corresponding spectrum in DMSO-d_6 . Thus, the sodium salt of 3 also exists preferentially in boat-chair conformation in D_2O solution.

This is the first case of a boat-chair conformational preference of a bicyclo{3.3.1}nonan-9-one due to the presence of exo-substituents at C_2 and C_4 . Work is in progress to prepare related compounds in which both the chair-chair and the boat-chair conformations are nearly equally populated.

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